(7 pages)

Reg. No. :

Code No.: 6526

# Sub. Code: ZCHM11

## M.Sc. (CBCS) DEGREE EXAMINATION, NOVEMBER 2021

First Semester

Chemistry - Core

## AROMATICITY AND ORGANIC REACTION MECHANISM

(For those who joined in July 2021 onwards)

Time : Three hours

Maximum : 75 marks

PART A —  $(10 \times 1 = 10 \text{ marks})$ 

Answer ALL questions.

Choose the correct answer:

- 1. Which of the following is a characteristic of an aromatic compound?
  - (a) Cyclic (b) Planar
  - (c)  $(4n+2)\pi$  electrons (d) All of the above

- 2. Which of the following is not a characteristic property of arenes?
  - (a) Delocalisation of  $\pi$  -electrons
  - (b) Resonance
  - (c) Greater stability
  - (d) Electrophilic additions
- 3. The number of transition states in a two step reaction is \_\_\_\_\_.
  - (a) None (b) 1
  - (c) 2 (d) 4
- 4. Which pair of isotopes are likely to result in the greatest isotope effect?
  - (a) Carbon-12 and carbon-14
  - (b) Carbon-12 and carbon-13
  - (c) Hydrogen and deuterium
  - (d) Nitrogen-14 and nitrogen-15
- 5. Which types of isomers are formed in rearrangement reactions?
  - (a) Structural isomers
  - (b) Geometrical isomers
  - (c) Optical isomer
  - (d) Conformational isomers

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- 6. What is the main difference between Hofmann and Curtius rearrangement?
  - (a) Products are different
  - (b) Intermediate formed is different
  - (c) Reactants are different
  - (d) Isomers
- 7. Which of the following statements regarding the EI mechanism is wrong?
  - (a) Reactions by the E1 mechanism are unimolecular in the rate-determining step.
  - (b) Reactions by the E1 mechanism are generally first order.
  - (c) Reactions by the E1 mechanism usually occur in one step.
  - (d) Reactions by the E1 mechanism are multistep reactions.
- 8. Which of the following statements regarding the E2 mechanism is wrong?
  - (a) Reactions by the E2 mechanism are always bimolecular.
  - (b) Reactions by the E2 mechanism are generally second order.
  - (c) Reactions by the E2 mechanism usually occur in one step.
  - (d) Reactions by the E2 mechanism usually occur in two steps.

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- 9. The compound below that would react fastest in a nucleophilic aromatic substitution reaction is
  - (a) p-nitrobromobenzene
  - (b) m-nitrohromobenzene
  - (c) 2,4-dinitrohromobenzene
  - (d) 3.4-dinitrobromobenzene

10. Using the benzyne mechanism for substitution of m-chlorotoluene by  $NaNH_2$ . the possible products are

- (a) o-methylaniline (b) m-methylaniline
- (c) p-methylaniline (d) All of the above

PART B —  $(5 \times 5 = 25 \text{ marks})$ 

Answer ALL questions, choosing either (a) or (b). Each answer should not exceed 250 words.

11. (a) What are the differences between beuzenoid and non- benzenoid compounds?

 $\mathbf{Or}$ 

(b) Explain delocalization and resonance.

Page 4 Code No. : 6526 [P.T.O.] 12. (a) Explain Hammond postulates.

Or

- (b) Explain principle of microscopic reversibility.
- (a) What is Wolff rearrangement? Explain the mechanism of Wolff rearrangement.

### Or

- (b) What is Schmidt rearrangement? Explain the mechanism of Schmidt rearrangement.
- 14. (a) Explain the mechanism of  $S_N 1$  reaction with an example.

#### Or

- (b) Explain the mechanism of  $S_N 2$  reaction with an example.
- 15. (a) Explain benzyne mechanism with example.

#### $\mathbf{Or}$

- (b) What is Bucherer Reaction? Explain the mechanism of Bucherer Reaction.
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PART C —  $(5 \times 8 = 40 \text{ marks})$ 

- Answer ALL questions, choosing either (a) or (b) Each answer should not exceed 600 words.
- 16. (a) What is Frost Musulin diagram? How does aromaticity affect the NMR spectra?

Or

- (b) Explain aromaticity of annulenes and heteroannulenes. Explain Anternant and nonalternant hydrocarbons.
- 17. (a) Explain primary and secondary kinetic isotopic effects.

 $\mathbf{Or}$ 

- (b) Explain the significance of Hammett equation.
- 18. (a) What is Beckmann rearrangement? Explain the mechanism of Beckmann rearrangement. Mention the applications of Beckmann rearrangement.

 $\mathbf{Or}$ 

(b) What is Hoffmann rearrangement? Explain the mechanism of Hoffmann rearrangement. Mention the applications of Hoffmann rearrangement.

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19. (a) Explain the mechanism of  $S_N i$  reaction with an example.

Or

- (b) Write the mechanism of E1CB reaction with example. Compare Hoffmann elimination and Saytzeff elimination reactions.
- 20. (a) What is Smiles rearrangement? Discuss the mechanism of Smiles rearrangement.

Or

(b) What is Mannich reaction? Write the mechanism of Mannich reaction.

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