

(6 pages)

Reg. No. :

Code No. : 9387

Sub. Code : HPHM 42

M.Sc. (CBCS) DEGREE EXAMINATION, APRIL 2015.

Fourth Semester

Physics

MOLECULAR SPECTROSCOPY

(For those who joined in July 2012 onwards)

Time : Three hours

Maximum : 75 marks

SECTION A — (10 × 1 = 10 marks)

Answer ALL the questions.

Choose the correct answer :

1. For linear molecules

- (a) $I_B = I_C; I_A = 0$ (b) $I_A = I_B; I_C = 0$
(c) $I_A = I_C; I_B = 0$ (d) $I_A = I_B = I_C = 0$

2. For a rigid diatomic molecule, the absorption spectrum lines are with a line spacing of

- (a) 4B (b) B
(c) 2B (d) 5B

3. Zero-point energy of a simple harmonic vibrator is

- (a) $E_0 = \frac{1}{2}tw$ (b) $E_0 = \frac{w}{t}$
(c) $E_0 = \frac{t}{w}$ (d) $E_0 = t^2w$

4. The selection rule for a rotating-vibrating linear molecule is

- (a) $\Delta v = 0$
(b) $\Delta v = \pm 1$
(c) $\Delta v = \pm 1$ and $\Delta J = 0, \pm 1$
(d) $\Delta J = 0, \pm 1$

5. Stoke's radiation is generally

- (a) More intense than antistoke's lines
(b) Less intense
(c) Less intense than antistoke's lines
(d) More intense

6. The selection rule for pure rotational Raman spectra of linear molecules

- (a) $\Delta J = 0$
(b) $\Delta J = \pm 1$
(c) $\Delta J = \pm 2$
(d) $\Delta J = 0$ or $\Delta J = \pm 2$ only

Page 2

Code No. : 9387



7. Multiplicity of a system is given by

- (a) $2S + 1$ (b) $2S$
(c) $2S - 1$ (d) $S - 1$

8. Franck-Condon principle explains the intensity of

- (a) Rotation spectra
(b) Vibrational spectra
(c) Electronic spectra
(d) Vibrational-electronic spectra

9. Nuclear magneton β_N is equal to

- (a) $\frac{eh}{4m_p\pi}$ (b) $\frac{h}{4m_p\pi}$
(c) $\frac{e}{4m_p\pi}$ (d) $\frac{eh}{m_p}$

10. The coupling constant refers to

- (a) Spin-lattice interaction
(b) Spin-spin interaction
(c) Lattice-lattice interaction
(d) Molecular interaction

Page 3

Code No. : 9387

SECTION B — ($5 \times 5 = 25$ marks)

Answer ALL the questions, choosing either (a) or (b).

Each answer should not exceed 250 words.

11. (a) Deduce the rotational constant for a rigid diatomic molecule.

Or

- (b) Explain briefly the rotational spectrum of asymmetric molecule.

12. (a) Discuss the vibrational energy of a diatomic molecule.

Or

- (b) Explain briefly the asymmetry of vibration.

13. (a) Give the quantum theory of Raman scattering.

Or

- (b) Give the classical treatment of hyper Raman effect.

14. (a) Explain the vibrational coarse structure.

Or

- (b) Explain the predissociation of spectral lines.

Page 4

Code No. : 9387

[P.T.O.]



15. (a) Derive the resonance condition for NMR.

Or

- (b) Explain the quadrupole nucleus.

SECTION C — ($5 \times 8 = 40$ marks)

Answer ALL the questions, choosing either (a) or (b).

Each answer should not exceed 600 words.

16. (a) Obtain the energies of a non-rigid rotator.

Or

- (b) Describe the information derived from rotational spectra.

17. (a) Obtain the energies of a diatomic vibrating rotator.

Or

- (b) Describe the rotational vibrational spectra of polyatomic molecules.

18. (a) Obtain the frequencies of Raman spectra of symmetric top molecules.

Or

- (b) Illustrate the vibrational Raman spectra with suitable example.

Page 5

Code No. : 9387

19. (a) Discuss the rotational fine structure of electronic vibrational spectra.

Or

- (b) Give an account of photoelectron spectroscopy.

20. (a) Explain the ESR spectrum of hydrogen atom.

Or

- (b) Give an account of NQR instrumentation.

Page 6

Code No. : 9387

